Chemical Composition and Antioxidant Properties of Ferula-assa-foetida Leaves Essential Oil

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ABSTRACT
Recent investigations have shown that plants have antioxidant properties and could decrease oxidative stress in different human diseases. The present work was identified the chemical composition compounds and evaluated the various antioxidative activities of Ferula-assa-foetida leaves hydroalcoholic extract (FLHE) and Ferula-assa-foetida leaves essential oil (FLEO). Radical scavenging activity of samples was assessed using Diphenylpicrylhydrazyl (DPPH). Total antioxidant capacity was assessed by methods phosphomolybd. The amount of total phenol and flavonoid was assessed by Folin-Ciocalteu and Zhishen methods. Also, the components of FLEO were analyzed with gas chromatography/mass spectrometry (GC/MS). In FLHE and FLEO, total antioxidant capacity were 1.55 ± 0.13 and 1.09 ± 0.43 nmol, ascorbic acid equivalents/g, phenol content were 441.37 ± 2.50 and 76.56 ± 9.65 mg of Gallic acid equivalents (GAE)/g, and flavonoid content was 12.53 ± 3.20 and 0.015 ± 0.002 mg of quercetin equivalents/100 g respectively. In the DPPH scavenging assay, the IC50 values of FLHE, FLEO and Butylated hydroxytoluene (BHT) as reference were 787.13 ± 3.66; 2375.66 ± 5.13 and 3.88 ± 1 g/mL respectively. Also GC/MS data and retention indices for reference essential oil leaves samples were used to identify 12 constituents. These compounds made up a total of 85.57% of FLEO: Eremophilene; δ-cadinene; Longiborneol; Dehydro aromadendrene; δ-isoleđene; γ-Gurjunene; δ-Guaiene. This study showed that ferula-assa-foetida has good antioxidant properties. As it is an easily-accessible source of natural antioxidants such as Eremophilene and δ-cadinene, it may be suitable for use in food and pharmaceutical applications.

Keywords: Total phenols, Total flavonoids, Antioxidant activity, Radical scavenging activity, DPPH; Essential oil, Chemical composition, Ferula-assa-foetida

Oxidative stress is an imbalance between productions of free radicals and the body’s antioxidant defense mechanisms. Lipid peroxidation in the walls of living cells is a marker of free radicals attack. The presence of peroxides, especially free radicals play a key role in the number of virulent diseases such as diabetes, heart disease, cancer, aging and other diseases [1,2]. Hence the agents that can scavenge these reactive oxygen species can be beneficial in the treatment of such inflammatory disorders.

Ferula assa-foetida grows in Afghanistan and Kashmir. Also ferula assa-foetida grows wildly in the central and southern mountains of Iran [3,4]. It is traditionally used for the treatment of different diseases, such as asthma, epilepsy, stomachache, flatulence, intestinal parasites, weak digestion and influenza [4,5]. Recent pharmacological and biological studies have also shown several activities, such as antioxidant [3], antiviral, antifungal, cancer chemopreventive [6], anti-diabetic, and antispasmodic and hypotensive from Ferula assa-foetida [4,7]. Ferula-assa-foetida has good effects as herbal drug. These effects are related to chemical composition of Ferula-assa-foetida. Since the various antioxidative activities of Lorestan FLHE and
Ferula and chemical composition compounds of Lorestan ferula-assafoetida leaves essential oil have not previously been reported; the objective of the present study is to investigate various antioxidative activities of Lorestan FLHE and FLEO and chemical composition compounds of Lorestan FLEO.

**MATERIALS AND METHODS**

**Isolation of the Hydroalcoholic extract and essential oil from Ferula-assafoetida leaves**

Ferula-assafoetida were prepared in July 2012 from cultivated plant in Khoram Abad (Lorestan province, western Iran). Leaves of the plants were collected during flowering stage and were air-dried at ambient temperature in the shade separately. The voucher specimen (no. 1209) was deposited at Herbarium of the Agriculture and Natural Resources Research Center of Lorestan Province, Khoramabad, Iran. Leaves were hydro-distilled using a Clevenger apparatus for 4 h, giving yellow oil in 0.38% yield. The oil was dried over anhydrous sodium sulfate and stored at 4ºC. Also, hydroalcoholic extract of leaves at the Research Center of Lorestan University of Medical Sciences was prepared.

**DPPH free radical-scavenging activity**

DPPH free radical-scavenging activity of the test samples was determined according to the method of Blois (1958) [8]. In brief, 4 mL of DPPH radical solution in ethanol (1 mM) was mixed with 1 mL of Ferula-assafoetida extract or essential oil solution in ethanol containing 0.01–1000 µg/mL of FLHE or FLEO; and after 30 min, the absorbance was measured at 517 nm. This activity was given as percentage DPPH scavenging that is calculated as %DPPH scavenging = [(control absorbance - FLHE or FLEO absorbance) / (control absorbance)] ×100. The 50% inhibition concentration (IC50), i.e., the concentration of FLHE or FLEO that was required to scavenge 50% of radicals, was calculated [8].

**Total antioxidant activity**

Total antioxidant activity of the test samples was determined according to a previously-described procedure [9]. In brief, 0.3 mL of sample was mixed with 3.0 mL of reagent solution (0.6 M sulfuric acid, 28 mM sodium phosphate and 4 mM ammonium molybdate). Reaction mixture was incubated at 95ºC for 90 min under water bath. Absorbance of all the sample mixtures was measured at 695 nm. The total antioxidant activity was expressed as the number of equivalents of ascorbic acid (mol g⁻¹).

**Total phenolic content measurement**

Total phenols content in the obtained extracts was determined using the Folin–Ciocalteu’s phenol reagent, according to a previously-described procedure [10], with some modifications. Briefly, 200 L of the extract solution was mixed with 1.5 mL of Folin-Ciocalteau reagent (previously diluted 10-fold with distilled water). After 3 min, 1.5 mL sodium bicarbonate solution (60 g L⁻¹) was added to the mixture. The reaction was kept in the dark for 90 min, after which the absorbance was read at 725 nm (Analytic Jena 200-2004 spectrophotometer). Gallic acid was used for constructing the standard curve (1–3500 µg/mL). The results were expressed as mg of Gallic acid equivalents (GAE)/g of extract.

**Determination of total flavonoids**

Total flavonoid contents were determined using the method of Ordon ez et al. (2006) [11]. A volume of 0.5 mL of 2% AlCl3 ethanol solution was added to 0.5 mL of sample solution. After one hour at room temperature, the absorbance was measured at 420 nm. A yellow color indicated the presence of flavonoids. Quercetin was used for constructing the standard curve (1–50 µg/mL). The results were expressed as mg of quercetin equivalents /g of extract [11].

**Gas chromatography/ mass spectrometry**

Fid- GC was carried out using a Hewlett-Packard 6890 with HP-5 capillary column (phenyl methyl silicone, 25 m, 0.25 mm i.d., ratio, 1:25, and flame ionization detector. Temperature programme: 60ºC (2 min) rising to 240ºC at 4ºC/min; injector temperature 250ºC, detector temperature, 260ºC. GC-MS was performed using a Hewlett-Packard 6859 with a quadrupole detector, on a HP-5 column, operating at 70 eV ionization energy, using the same temperature programme and carrier gas as above. Retention indices were calculated by using retention times of n-alkanes that were injected after the oil at the same chromatographic to Van Den Dool method.

**Identification of components**

The linear retention indices for all the compounds were determined by conjection of the sample with a solution containing the homologous series of C8-C22 n-alkanes. The individual constituents were identified by their identical retention indices, referring to known compounds from the literature and also by comparing their mass spectra with either the known compounds or with the Wiley mass spectral database.

**Statistical analysis**

The data were presented as mean ± SD of three experiments performed in duplicate. These parameters were obtained using the student’s t-test (using SPSS 13.0 statistical software) for independent data and the differences were considered significant when p < 0.05.

**RESULTS AND DISCUSSION**

The polyphenol compounds exhibit a wide variety of beneficial activities in mammals including anti-diabetic [12], anti-bacterial [13], immune stimulating [14], anti-allergic [15], anti-hypertensive [16], anti-ischemic,
anti-arrhythmic [17], anti-thrombotic [18], hypocholesteromic, hepatoprotective [19], and anti-inflammatory [20], anticarcinogenic [21,22]. Flavonoids are an important group of polyphenols with antioxidant, anti-viral, anti-bacterial [23] and anti-inflammatory activities [20,24]. The present investigations have demonstrated a strong correlation between antioxidant activities and chemical composition content of *Ferula assa-foetida* leaves. A stable free radical 2, 2-diphenyl-1-pycrylhydrazyl (DPPH) has widely been used in the assessment of radical scavenging activity of plant extracts, natural compounds and foods [25].

**DPPH radical scavenging capacity**

The antioxidant activity of *Ferula assa-foetida* leaves was evaluated by the DPPH radical scavenging capacity. Table 1 shows the percentage of DPPH radicals scavenging capacity FLHE, FLEO and BHT as reference. In the DPPH scavenging assay, the IC$_{50}$ (the concentration required to scavenge 50% of radical) values of FLHE, FLEO and BHT as reference were 787.13 ± 3.66; 2375.66 ± 5.13, 3.88 ± 1 µg/ml, respectively. The data obtained show that *Ferula assa-foetida* leaves is free radical scavenger and may act as primary antioxidant, which can react with free radicals by donating hydrogen. Nabavi et al. (2011) reported the IC$_{50}$ values of *ferula-assa-foetida* leaves ethanol/water (70:30; V/V) extract prepared from cultivated in Gadouk area (central Elburz, Iran) was 1925 ± 8.6 µg/ml [3]. Also Zia-UL-Haq et al. (2012) reported the IC$_{50}$ values of *ferula-assa-foetida* leaves aqueous methanolic mixture (80:20; V/V) extract prepared from Pakistan was 99.36 ± 0.28 µg/ml [26].

Our findings shows that the anti-oxidant properties of *ferula-assa-foetida* leaves ethanol/water (50:50; V/V) extract are higher than those of *ferula-assa-foetida* leaves ethanol/water (70:30; V/V) extract and less than those of *ferula-assa-foetida* leaves aqueous methanolic mixture (80:20; V/V) extract. Researchers showed the presence of different flavonoids, phenolic compound, Eremophilene and δ-cadinene in *ferula-assa-foetida* leaves [27]. These compounds may be the main cause of its considerable radical-scavenging activity.

Researchers are recently get interested in investigation and research into extraction of natural antioxidants such as coenzyme Q10, rosmarinic acid, tannins and flavonoids, from medical herbs to replace synthetic antioxidants [2,28]. Natural antioxidants are safer and more beneficial and have fewer side effects than synthetic antioxidants [29]. Phytochemical with antioxidant effects include some cinnamic acids, oleuropein, coumarins, falvonoids, eremophilene and tannins [30,31]. Therefore, the herbs which have high amount of these compounds are taken into consideration in order to inhibit diseases related to oxidative stress such as coronary heart disease and diabetes [32]. Conducting research on herbal antioxidants and evaluating and comparing their antioxidant effects, as well as newer and more valuable sources of natural antioxidants can be found and used in food and pharmaceutical applications.

**Total antioxidant activity**

The phosphomolybdenum method has been widely used in the assessment of total antioxidant activity of plant extracts, natural compounds and foods. The total antioxidant activity of FLHE and FLEO were 1.55±0.13 and 1.09±0.43 nmol of ascorbic acid equivalents/g respectively (Table 2). The difference in the amount of antioxidant of extracts may be attributed to the

<table>
<thead>
<tr>
<th>Parameter</th>
<th>FLHE</th>
<th>FLEO</th>
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<tbody>
<tr>
<td>Total phenols (nmol ascorbic acid equivalents/g FLHE or FLEO)</td>
<td>1.55 ± 0.13*</td>
<td>1.09 ± 0.43</td>
</tr>
<tr>
<td>Total flavonoids (nmol of quercetin equivalents/100g FLHE or FLEO)</td>
<td>441.37 ± 12.03*</td>
<td>76.56 ± 9.65</td>
</tr>
<tr>
<td>Abbreviations as in Table 1. Each point represents the mean of five experiments</td>
<td></td>
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</table>

*$p<0.05$ as compared with FLEO

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**Table 1.** The IC$_{50}$ (the concentration required to scavenge 50% of radical) values of *Ferula assa-foetida* leaves hydroalcoholic extract (FLHE), *Ferula assa-foetida* leaves essential oil (FLEO) and BHT. Each point represents the mean of five experiments.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Mean</th>
<th>S.D.</th>
</tr>
</thead>
<tbody>
<tr>
<td>FLHE (µg/ml)</td>
<td>787.13*</td>
<td>3.66</td>
</tr>
<tr>
<td>FLEO</td>
<td>2375.66</td>
<td>5.13</td>
</tr>
<tr>
<td>BHT (µg/ml)</td>
<td>3.88*</td>
<td>1.00</td>
</tr>
</tbody>
</table>

* $p<0.05$ as compared with FLEO.
differences in the amount and kind of existing antioxidant compounds.

**Total phenols**

The total phenol of FLHE and FLEO were 441.37 ± 12.03 and 76.56 ± 9.65 of Gallic acid equivalents/g respectively (Table 2). Total phenol shown by the FLHE or FLEO may be due to the presence of different phenolic compounds. Nabavi et al. (2011) reported the total phenol of ferula-assa-foetida leaves ethanol/water (70/30; V/V) extract prepared from cultivated in Gadouk area (central Elburz, Iran) was 49.4 ± 2.1 of Gallic acid equivalents/g extract [23]. Also Zia-UL-Haq et al. (2012) reported the total phenol of ferula-assa-foetida leaves aqueous methanolic mixture (central: 20; V/V) extract was 437.67 ± 1.44 of Gallic acid equivalents/g extract [26]. Our findings on the total phenol of ferula-assa-foetida leaves ethanol/water (50:50; V/V) extract was higher than that of ferula-assa-foetida leaves ethanol/water (70:30; V/V) extract and less than that of ferula-assa-foetida leaves aqueous methanolic mixture (80:20; V/V) extract. Our recent results indicated that ferula-assa-foetida leaves extract possess a good antioxidant activity.

**Total flavonoids**

The total flavonoid of FLHE and FLEO were 12.53 ± 3.2 and 0.015 ± 0.002 mg of quercetin equivalents/g respectively (Table 2). Total flavonoid shown by the FLHE or FLEO may be due to the presence of different phenolic compounds.

### Table 3. Chemical composition of the Ferula-assa-foetida leaves essential oil (FLEO)

<table>
<thead>
<tr>
<th>Compound</th>
<th>Retention Time (min)</th>
<th>Yield %</th>
</tr>
</thead>
<tbody>
<tr>
<td>Eremophilene</td>
<td>39.35</td>
<td>31.28</td>
</tr>
<tr>
<td>δ-cadinene</td>
<td>37.67</td>
<td>22.05</td>
</tr>
<tr>
<td>Longiborneol</td>
<td>33.90</td>
<td>12.09</td>
</tr>
<tr>
<td>Dehydromadendrene</td>
<td>39.70</td>
<td>3.99</td>
</tr>
<tr>
<td>Isoledene</td>
<td>36.43</td>
<td>3.98</td>
</tr>
<tr>
<td>τ-Gurjunene</td>
<td>35.83</td>
<td>3.93</td>
</tr>
<tr>
<td>J-Guaiene</td>
<td>38.11</td>
<td>3.53</td>
</tr>
<tr>
<td>α-pinene</td>
<td>6.08</td>
<td>1.47</td>
</tr>
<tr>
<td>Ledenoxid</td>
<td>34.77</td>
<td>1.04</td>
</tr>
<tr>
<td>Trans-caryophyllene</td>
<td>22.70</td>
<td>0.98</td>
</tr>
<tr>
<td>α-Gurjunene</td>
<td>37.98</td>
<td>0.95</td>
</tr>
<tr>
<td>2-J-pinene</td>
<td>7.39</td>
<td>0.23</td>
</tr>
</tbody>
</table>

### Table 4. Bioactive and fragrance components percentage of some Ferula species essential oils

<table>
<thead>
<tr>
<th>Plant Name [Ref.]</th>
<th>Major Components (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>F. szovitsiana leaf [19]</td>
<td>β-Eudesmol (32.0), α-Eudesmol (18.2), α-Pinene (8.6)</td>
</tr>
<tr>
<td>F. arrjjonii leaf [28]</td>
<td>β-Phellandrene (16.6), Germacr-1 (10),5-dien,11-ol- (15.4)</td>
</tr>
<tr>
<td>F. communis leaf [29]</td>
<td>Myrcene (53.5), Aristolene (8.5)</td>
</tr>
<tr>
<td>ferula-assa-foetida leaves [our study]</td>
<td>Eremophilene (31.28%) and δ-cadinene (22.05)</td>
</tr>
</tbody>
</table>

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flavonoids such as catechin, epicatechin and myricetin. Nabavi et al. (2011) reported the total flavonoid of *ferula-assa-foetida* leaves ethanol/water (70:30; V/V) extract prepared from cultivated in Gadouk area (central Elburz, Iran) was 20.9 ± 1.4 mg of quercetin equivalents/g extract [3]. Also, Zia-Ul-Haq et al. (2012) reported the total flavonoid of *ferula-assa-foetida* leaves aqueous methanolic mixture (80:20; V/V) extract prepared from Pakistan was 416.3± 1.6 mg of quercetin equivalents/g extract [26]. Our findings on the total flavonoid of *ferula-assa-foetida* leaves ethanol/water (50:50; V/V) extract are less than that of *ferula-assa-foetida* leaves ethanol/water (70:30) extracts and *ferula-assa-foetida* leaves aqueous methanolic mixture (80:20; V/V) extract. Our recent results indicated that FLHE or FLEO possess a good antioxidant activity.

**Chemical composition of ferula-assa-foetida leaves essential oil**

The yield of the essential oils obtained from leaves was 0.8% (W/W) respectively. Results of the GC-MS analysis of the oils are shown in Table 3. Twelve compounds of FLEO were identified (85.57% of the total oils respectively).

The main constituents found in the *Ferula-assa-foetida* leaves were Eremophilene (31.28%); δ-cadinene (22.05%); Longiborneol (12.09%); Dehydro aromadendrene (3.99%); Isolatedene (3.98%); γ-Gurjunene (3.93%); J-Guaiene (3.53%); α-pinene (1.47%); Lenedoxoid (1.04%); Trans-caryophyllene (0.98%); α-Gurjunene (0.95%); 2-J-pinene (0.23%).

Kose et al. (2010) reported, the main constituents found in the *Ferula-assa-foetida* aerial parts prepared from Turkey were α-Pine (59.89%); β-Pinene (13.48%); Limonene (3.21%) and Bornyl acetate (2.10%) [33]. Also, another study reported, the main constituents of *Ferula-assa-foetida* aerial parts prepared from Sari forest (mazandaran, Iran) were Phenol, 2-methyl-5-(1-methylethyl) (18.2%); Bisabolol (10.4%); Fenchyl acetate(4.7%); γ-Gurjunene (2.3%) and α-Pinene (0.6%) [27]. Also, according to a similar experiment on the essential oil from of *F. assa-foetida* leaves in the province of Kerman, south of Iran, in August 2005 [34], the major components were reported as E-1-Propenyl sec-butyl disulfide (40.0%) and Germacrene B (7.8%), while in our work, Eremophilene (31.28%) and δ-cadinene (22.05%) are presented as major components in oils from leaves in Lorestan species, respectively. As seen, can be the chemical composition of *F. assa-foetida* in our results is different from the other one. These differences between the oil compositions of one species of *Ferula* from two different areas can be related to climatological factors (from south to central Iran) and is discussed in phytophology studies. Previous studies on essential oils of members of *Ferula* genus showed various compositions. Table 4 shows the diversity of major components among oils of some *Ferula* species which have been analyzed before [34-37]. β-Eudesmol (32.0), α-Eudesmol (18.2), α-Pinene (8.6) in *F. szowitsiana leaves* [35], β-Phellandrene (16.6), Germacra-1(10)-dien,11-ol- (15.4) in *F. arrijonii* [36] and Myrcene (53.5), Aristolene (8.5) in *F. communis* [37] oils are the major components.

The results of this study indicate that the composition of volatile oil of *Ferula-assa-foetida* leaves is not similar to those which are reported in other studies. The observed differences may be probably due to using of different parts of plant for analysis, different environmental and genetic factors, different chemotypes and the nutritional status of the plants as well as other factors that can influence the oil composition.

In conclusion, This study showed that essential oil of *ferula-assa-foetida* leaf is an easily-accessible source of natural antioxidants such as eremophilene, δ-cadinene and longiborneol and it may be suitable for use in food and pharmaceutical applications and a good alternative to reduce the risk of atherosclerosis and coronary heart disease and other free radical-associated health problems.

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**References**

Chemical composition of Ferula assafoetida


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